# Synthesis and Crystal Structure of (2E)-1-(Anthracen-9-yl)-3-(3,4-dichlorophenyl)prop-2-en-1-one 

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#### Abstract

E)-1-(Anthracen-9-yl)-3-(3,4-dichlorophenyl)prop-2-en-1-one. $\quad C_{23} H_{14} \mathrm{OCl}_{2}$, crystallizes in the monoclinic space group $P 2 / c$ with unit cell parameters, $a=13.7669(9) \AA, b=11.8400(10) \AA$ À, $c=11.3285(8)$ $\AA, \beta=103.525(6)^{\circ}$ and number of molecules per unit cell $(Z)=4$. The crystal structure was solved by direct methods and refined by full matrix least square procedure to a final $R$-value of 0.067 for 2233 observed reflections. The dihedral angle between the benzene and anthracene moiety of the molecule is $89.43(1)^{\circ}$.The structure is stabilized by two $C-H \cdots O$ intermolecular interaction. Aromatic $\pi-\pi$ stacking interaction has also been observed in the structure.


Keywords: X-ray Structure, Intermolecular Interactions, Direct Methods, Graph Set Motif.
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## I. Introduction

Chalcones are an important class of natural compounds and have been widely applied as synthons in synthetic organic chemistry. Naturally occurring chalcones as well as their synthetic analogues have demonstrated interesting biological activities such as, anticancer[1]', anti-invasive[2], anti-tuberculosis[3], antimicrobial[4], anti-malarial[5], antitumor[6], antiproliferative[7] and antioxidant activity[8]. The nonlinear optical [NLO] properties of the different chalcone derivatives have also been reported[9-12]. These $\alpha$, $\beta$ unsaturated ketones are biosynthetic precursors of various functionalized derivatives[13], pyrazolines[14] and triaryl pyridines[15].In view of the importance of chalcones derivatives and our past published work[16-18].The synthesis and crystallographic analysis of (2E)-1-(Anthracen-9-yl)-3-(3,4-dichlorophenyl)prop-2-en-1-one is reported in this paper.

## II. Experimental

### 2.1 Synthesis of (2E)-1-(Anthracen-9-yl)-3-(3,4-dichlorophenyl)prop-2-en-1-one.

In 9-acetylanthracene ( 0.01 mol ) and 3,4-dichlorobenzaldehyde ( 0.01 mol ) in ethanol ( 50 mL ), 15 mL of $10 \%$ sodium hydroxide solution was added and stirred at $0-5{ }^{\circ} \mathrm{C}$ for 3 h . The precipitate formed was collected by filtration and purified by recrystallization from ethanol. Single crystal was grown from DMF by slow evaporation method (M.P.: 449-451 K). The synthetic route of the compound is shown in Fig. 1.

### 2.2 X-ray data collection and structure refinement.

A well defined crystal of dimensions $0.30 \times 0.20 \times 0.10 \mathrm{~mm}^{3}$ was used for data collection on X'calibur CCD area-detector diffractometer equipped with graphite monochromated MoK $\alpha$ radiation ( $\lambda=0.71073 \AA$ ). Xray intensity data of 7121 reflections were collected at $293(2) \mathrm{K}$ and out of these reflections 3509 were found unique. The intensities were measured by $\omega$ scan mode for $\theta$ ranges $3.61^{\circ}$ to $26.00^{\circ} .2233$ reflections were treated as observed using ( $\mathrm{I}>2 \sigma(\mathrm{I})$ ) as criterion. Data were corrected for Lorentz-polarization and absorption factors. The structure was solved by direct methods using SHELXS97[19] software. All non-hydrogen atoms of the molecule were located from the best E-map and the hydrogen atoms were geometrically fixed and allowed to ride on the corresponding non -H atoms with $\mathrm{C}-\mathrm{H}=0.93 \AA$ and $\mathrm{U}_{\mathrm{iso}}=1.2 \mathrm{U}_{\mathrm{eq}}(\mathrm{C})$. The final refinement cycles converged to an R - index of $0.0674(\mathrm{wR}(\mathrm{F} 2)=0.1831)$ for the 2233 observed reflections. Residual electron densities ranges from -0.446 to $0.875 \mathrm{e}^{\circ} \AA^{-3}$. Atomic scattering factors were taken from International Tables for Xray Crystallography. Allied structural calculations of the molecule were done using the WinGX [20], PARST [21] and PLATON [22] softwares.

Crystallographic information has been deposited to Cambridge Crystallographic Data Centre with CCDC number 1508193. This data can be obtained free of charge at Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

## III. Results And Discussion

The molecular structure containing atomic labeling is shown in Fig. 2 (ORTEP)[23] and the packing diagram is shown in Fig. 3 (PLATION)[22]. The molecule consists of a benzene and anthracene moiety. The structural parameters, including bond distances and angles of the molecule lie within the normal range [24] and are in good agreement with some related structures [25]. The anthracene and benzene rings are essentially planar with maximum deviation ( 0.0576 ) $\AA$ observed for C 8 [anthracene ring] atom and $(-0.0049) \AA$ corresponding to C 18 [benzene ring] atom. The double bonds $\mathrm{C} 16=\mathrm{C} 17$ and $\mathrm{C} 15=\mathrm{O}$ are confirmed by their respective distances of $1.321(5) \AA$ and $1.218(4) \AA$, respectively. These values are also consistent with corresponding ones observed in some related structures[25-27]. The anthracene ring system ( $\mathrm{C} 1-\mathrm{C} 14$ ) is twisted at the $\mathrm{C} 11-\mathrm{C} 15$ bond from the (2E)-1-(Anthracen-9-yl)-3-(3,4-dichlorophenyl)acrylaldehyde moiety with the torsion angle [C12-C11-C15-C16] of 93.4(4). The benzene and anthracene moiety (C18-C23 and C1-C14 respectively) forms a dihedral angle of $89.43(1)^{\circ}$ which makes them inclined at right angles to each other. The oxygen (O) atom attached to carbon atom (C15) adopts the + anti-clinical and - syn-clinical conformation with torsion angles (O-C15-C11-C10) 91.0(4) ${ }^{\circ}$ and (O-C15-C11-C12) -86.4(4) ${ }^{\circ}$ respectively. Molecular packing in the unit cell viewed down the $b$-axis is shown in Fig. 3. The structure is stabilized by the two $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ intermolecular interactions, where ' O ' act as a bifurcated acceptor to C 17 (via H17) and C 23 (via H23) atoms (Fig 4.). In the molecular packing, the adjacent molecules are interconnected through $\mathrm{C} 17-\mathrm{H} 17 \cdots \mathrm{O}$ and $\mathrm{C} 23-\mathrm{H} 23 \cdots \mathrm{O}$ hydrogen bond and $\pi-\pi$ interactions. Details of intermolecular hydrogen bonding and $\pi-\pi$ interactions are given in Table 3 and 4 , respectively.

## IV. Figures And Tables





Figure 1. Synthesis of (2E)-1-(Anthracen-9-yl)-3-(3,4-dichlorophenyl)prop-2-en-1-one.


Figure 2. ORTEP view of molecules with displacement ellipsoids drawn at $40 \%$ probability level. H atoms are shown as small spheres of arbitrary radii.


Figure 3. Packing viewed down the b-axis.


Figure 4. View of bifurcated (acceptor) hydrogen bond and $R_{2}^{1}(6)$ ring motif .
Table 1. Crystal data and other experimental details.

| CCDC Number | 1508193 |
| :--- | :--- |
| Crystal size | $0.30 \times 0.20 \times 0.10 \mathrm{~mm}$ |
| Empirical formula | $\mathrm{C}_{23} \mathrm{H}_{14} \mathrm{Cl} \mathrm{C}_{2} \mathrm{O}$ |
| Formula weight | 377.24 |
| Radiation, wavelength | $\mathrm{MoKa}, 0.71073$ |
| Unit cell dimensions | $\mathrm{a}=13.7669(9) \AA$, |
|  | $\mathrm{b}=11.8400(10) \AA$, |
|  | $\mathrm{c}=11.3285(8) \AA$. |
| Crystal system | Monoclinic |
| Space group | $\mathrm{P} 2_{1} / \mathrm{c}$ |
| Unit cell volume | $1795.3(2) \AA^{3}$ |
| No. of molecule per unit cell, Z | 4 |
| Absorption coefficient | $0.370 \mathrm{~mm}^{-1}$ |
| $\mathrm{~F}(000)$ | 776 |

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| $\Theta$ range for entire data collection | $3.61<\theta<26.00$ |
| :--- | :--- |
| Range of indices | $-12 \leq \mathrm{h} \leq 16$ |
|  | $-12 \leq \mathrm{k} \leq 14$ |
|  | $-13 \leq 1 \leq 12$ |
| Reflection collected/ unique | $7121 / 3509$ |
| Reflection observed (I $>2 \sigma(\mathrm{I})$ ) | 2233 |
| R int | 0.0313 |
| R sigma | 0.0567 |
| No. of parameter refined | 235 |
| Final R | 0.0674 |
| wR(F2) | 0.1831 |
| Goodness-of-fit | 1.042 |
| Final residual electron density | -0.446 to $0.875 \mathrm{eA}^{-3}$ |

Table 2. Selected bond lengths $(\AA)$, bond angles $\left({ }^{\circ}\right)$ and torsion angles $\left({ }^{\circ}\right)$ for non hydrogen atoms (e.s.d.'s are given in parentheses)

| Bond lengths |  | Bond angles |  | Torsion angles |  |
| :--- | :--- | :--- | :--- | :--- | :--- |
| O-C15 | $1.218(4)$ | C12-C11-C15 | $119.1(3)$ | O-C15-C11-C12 | $-86.4(4)$ |
| C11-C15 | $1.514(5)$ | C15-C10-C11 | $119.4(3)$ | C11-C15-C16-C17 | $-6.1(6)$ |
| C17-C18 | $1.460(5)$ | C11-C12-C13 | $123.4(4)$ | C16-C17-C18-C19 | $6.2(6)$ |
| C15-C16 | $1.447(6)$ | O-C15-C16 | $121.7(4)$ | C16-C17-C18-C23 | $-174.6(3)$ |
| C16-C17 | $1.321(5)$ | O-C15-C11 | $118.3(4)$ | C18-C23-C22-CL2 | $179.9(3)$ |
| C11-C10 | $1.402(5)$ | C22-C21-CL1 | $120.6(4)$ | C23-C22-C21-CL1 | $178.7(3)$ |
| C21-CL1 | $1.726(4)$ | C23-C22-CL2 | $118.8(3)$ | O-C15-C11-C10 | $91.0(4)$ |
| C22-CL2 | $1.700(4)$ | C20-C21-CL1 | $119.8(3)$ | C16-C15-C11-C10 | $-89.2(4)$ |
| C18-C23 | $1.394(5)$ | C21-C22-CL2 | $121.4(3)$ | CL1-C21-C20-C19 | $-179.1(3)$ |

Table 3. Hydrogen bonding geometry ( e.s.d.'s in parentheses).

| $\mathbf{D}-\mathbf{H} \ldots \mathbf{A}$ | $\mathbf{D}-\mathbf{H}(\AA)$ | $\mathbf{A})$ | $\mathbf{H} \ldots \mathbf{A}(\AA)$ | $\mathbf{D} \ldots \mathbf{A}(\boldsymbol{\AA})$ |
| :--- | :---: | :---: | :---: | :---: |
| $\mathrm{C} 17-\mathrm{H} 17 \ldots \mathrm{O}^{\mathrm{i}}$ | 0.93 | 2.55 | $3.410(4)$ | $\left.\mathbf{D}-\mathbf{H} \ldots \mathbf{A} \mathbf{o}^{\mathbf{o}}\right)$ |
| $\mathrm{C} 23-\mathrm{H} 23 \ldots \mathrm{O}^{\mathrm{i}}$ | 0.93 | 2.54 | $3.397(5)$ | 155 |

Symmetry code: (i) x,1/2-y.1/2+z.
Table 4. Geometry of $\pi-\pi$ interactions*

| CgI ...CgJ | CgI $\ldots \mathrm{CgJ}(\AA)$ | CgI...P $(\AA)$ | $\alpha\left({ }^{\circ}\right)$ | $\beta\left({ }^{\circ}\right)$ | $\Delta(\AA)$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Cg} 1 \ldots \mathrm{Cg} 3^{\mathrm{i}}$ | $3.8620(27)$ | 3.498 | 2.73 | 23.89 | 1.636 |
| $\mathrm{Cg} 2 \ldots \mathrm{Cg}{ }^{\mathrm{i}}$ | $3.8108(22)$ | 3.460 | 0.03 | 24.77 | 1.597 |
| $\mathrm{Cg} 3 \ldots \mathrm{Cg} 1^{\mathrm{i}}$ | $3.8620(27)$ | 3.531 | 2.73 | 25.09 | 1.564 |
| $\mathrm{Cg} 4 \ldots \mathrm{Cg}^{4 i}$ | $3.7446(23)$ | 3.487 | 0.02 | 21.39 | 1.364 |

Symmetry code: (i) $1-x,-y, 1-z$ (ii) $-x, 1-y, 1-z$

* $\mathrm{Cg} 1, \mathrm{Cg} 2, \mathrm{Cg} 3$ and Cg 4 represents the centre of gravity of the ring $\mathrm{A}, \mathrm{B}, \mathrm{C}$ and D .


## V. Conclusions

Synthesis of (2E)-1-(Anthracen-9-yl)-3-(3,4-dichlorophenyl)prop-2-en-1-one led to the single crystal grown from DMF by slow evaporation technique (M.P.: 449-451 K) and the molecular and crystal structure was determined using single crystal X-ray diffraction techniques. The structure was refined to final R-factor of 0.067 . The dihedral angle $\left[89.43(1)^{\circ}\right.$ ] between the benzene and anthracene moiety makes them held at right angles to each other. The structure is stabilized by few $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ and $\pi-\pi$ intermolecular interactions.

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