

Seed Characteristics and Proximate Analysis of Wild Castor Plant from Sokoto State

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Abstract: In this study, castor seeds were collected from Wamakko local government Area of Sokoto and the seeds were subjected to different studies which included seed characteristics, proximate analysis and mineral composition. The typical castor seed in Sokoto was found to have a range of 1- 8 bunch, each bunch contains several pods ranging from 30- 56 with an average of 43 pods per bunch. The proximate analysis of the seed was found to contain 28% carbohydrate, 11% protein, 3.5% ash, 1.78% nitrogen, 1.0 crude fibre respectively. The paper concludes that Sokoto state has good potential for large cultivation of castor plant and consequently joining the castor seed market.

Keywords; castor seeds, proximate analysis, seed characteristics, mineral composition

I. Introduction

The castor plant due to its unique characteristics is attracting a lot of interest as area of current research all over the world. Internationally, it is estimated that over 800,000 metric tons of castor seed are produced annually (Redmond, 2008). Primary exporters are China, Thailand, Russia and Sudan but Brazil and India grow significant quantities which are not traded internationally while the largest importers are the US, France and UK (Ahnet *et al.*, 2007). The global trade in this product is put at \$5b (Redmond, 2008). The castor bean contains 45-55% castor oil which makes it indispensable in tanning of skins, production of cosmetics, pharmaceuticals and insecticides (Ahnet *et al.*, 2007).

Castor seeds are nearly flattened and oval but differ in size¹ and colour (Fakhri, 1989). The size of castor seed has been reported to vary from a few millimetres long to about 2cm in the giant species (Das *et al.*, 2000). It has been reported that seeds from Nigeria and Kenya have an average weight of 59.2 and 61.3g per hundred seeds respectively (Fakhri, 1989). Small sized seeds weighing 16g per hundred have been reported in India (Willcox, 2000).

There are an astonishing numbers of industrial application for castor oil and its derivatives and new ones are continually been discovered (Raymond, 1996).

Sokoto state has a favourable climate for the large cultivation of castor plant as it is known to do well in the hot climate conditions and is drought tolerant (Iyothiet *et al.*, 2006). The castor plant has potential for export. In fact, the Yobe State government through the sponsorship of the EU recently embarked on a large scale production intended to engage about 100,000 farmers in the state. This effort is expected to boost the state economy as a major revenue generator and provider of job opportunities.

In Nigeria, no commercial castor oil is produced and the industries that need it depend on imported oil, which is very expensive (Das *et al.*, 2000). The need to promote the production of castor bean seed as an alternative to synthetic seed products is therefore beyond any emphasis.

II. Materials And Methods

Seed Collection and Handling

Castor fruit pods were collected between December 2009 and June, 2010 from Wamakko Local Government Area of Sokoto State on Longitude 5⁰ and Latitude 15⁰. The number of pods per plant was counted and sun dried to facilitate seed removal. The dried pods were then crushed and the number of seeds per pod was determined. The shelled seeds were cleaned by removing all debris, packed in large envelopes and kept in wooden cabinet.

Evaluation of Seed characteristics

The size, shape, colour, weight and density of castor seeds were evaluated in the laboratory. The length and width of ten seeds were determined using veneer calliper and the average taken. The colour was visually observed and the mass determined using a digital electronics balance (Citezen scales PVT LTD). The volume

of kerosene displaced by hundred castor seeds in a 50cm³ measuring cylinder was determined. The density was then evaluated using the mass and volume.

Analyses of Mineral composition in castor seeds

Dried castor seeds (1850g) were crushed to a paste using mortar and pestle. The castor seed paste was analysed to determine its mineral content. The minerals analysed were nitrogen, potassium, sodium, calcium and magnesium.

Determination of Nitrogen in the castor seed paste

In the determination of nitrogen Macro kjeldahl method was used.

Two grams of the sample were poured in a Macro kjeldahl flask and twenty ml of distilled water was added. The flask was swirled for a few minutes and allowed for 30 minutes to prevent foaming. One tablet of mercury was added, Fifty ml of concentrated H₂SO₄ was also added through an automatic pipette. The flask was heated cautiously at low temperature of 45°C on the digestion stand. When water has been removed and frothing has ceased, the temperature of the flask was increased until the digest was cleared. The digest was boiled for 5 hours. Heating was regulated during boiling so that H₂SO₄ condenser is about half way up the neck of the flask. The flask was allowed to cool and Fifty ml of distilled water was added to the flask. Ten ml of the aliquot was carefully transferred into a macro kjeldahl flask 750ml. Twenty ml H₃BO₃ indicator solution was added into Fifty ml Erlenmeyer flask which was then placed under the condenser of the distillation apparatus. Twenty ml of 40% NaOH was added to the macro kjeldahl flask through a funnel on the stop cork and distillation was commenced. The condenser was kept cool at 30°C allowing sufficient cold water to flow through and heat was regulated to minimize frothing and prevent suck back. 40ml distillate was collected and distillation was stopped. NH₄-N in the distillate was determined by titrating with 0.01N standard H₂SO₄ using burette graduated at 0.1ml intervals. The colour changed at the end point from green to pink. The percentage Nitrogen in the sample was calculated.

Thus –

$$\% \text{ nitrogen} = \frac{\text{TVX N} \times 0.014 \times \text{VOL.} \times 100}{\text{mass of sample} \times \text{aliquot of digest}}$$

Determination of Potassium

Potassium and sodium were determined by flame photometer and was set for potassium by inserting appropriate filter (usually of 788nm in wavelength). The instrument was set to 100% transmittance by feeding 10ppm K solution. All the standard solutions were run and standard curves prepared by plotting transmittance reading against concentration of standard potassium solution. The seed extract was run and the amount of potassium present in the sample per 100g oven dry mass of the sample was calculated by getting potassium concentration in the extract from standard curve. All dilutions were considered in making the calculations.

Determination of sodium

The flame photometer for sodium (Na) was set by inserting filter (usually of 589 nm wavelength). The instrument was set to 100% transmittance by feeding 10ppm sodium (Na) solution. The steps were repeated 3-5 times for potassium (K) determination.

Determination of calcium and magnesium

One millilitre aliquot of the extract was put into titration flasks using pipette and diluted to Twenty ml with distilled water. Five ml of buffer solution and 3 drops each of KCN, NH₂OH, HCl, K₄Fe [CN]₆ and triethanolamine were added. A few minutes was allowed for reaction to take place. Three drops of EBT indicator were added to the solution and titrated with (EDTA) in the above manner to the permanent blue colour. Calcium + magnesium were obtained by EDTA Titration method.

The Ca and Mg were obtained in the sample after making up to a definite volume. Aliquots of this extract were used to determine Ca +Mg and Ca alone. The value for Mg was obtained as the difference.

Proximate analysis of the castor seed paste

Proximate analysis of the castor seed paste was carried out in the Agriculture laboratory of Usmanu Danfodiyo University to determine the percentage of crude protein, lipid, ash moisture content, fibres, and carbohydrate of the seed paste.

Determination of the crude protein in the castor seed paste

The method used was the Macro kjeldahl method the principle was that when protein is boiled with concentrated sulphuric acid and a tablet of kjeldahl catalyst so the H₂SO₄, convert all forms of nitrogen to ammonia sulphate. Subsequently addition of an excess amount of NaOH in a close system neutralizes the acid

and releases ammonia which is distilled into boric acid solution, and titrated against 0.01N HCl or H₂OH₄ end point from green to pink.

Three steps were involved in this analysis. They are digestion, distillation and titration

Digestion

Two grams of the sample was weighed and placed in the bottom of a kjeldahl flask. Twenty (20) ml of concentrated H₂SO₄ was added to the flask which was swirled to soak the sample. One tablet of kjeldahl catalyst was added to it. The flask was heated gently on an electric heater in a fumes chamber until the solutions become blackened and then clear so as to convert any nitrogen present to ammonium sulphate and organic matter to carbon [iv] oxide.

Distillation and Titration

Ten mills of the digest sample were pipetted in to twenty mills of round bottom flask was heated in the macro kjeldahl apparatus for ten minutes. Twenty mills of boric acid indicator was placed in a 100ml conical flask and placed under the condenser Such that the condenser tip was above 4cm to the surface followed by Twenty (20) mills of 40% NaOH solution. The NaOH solution was let in carefully through the funnel and little was left behind to prevent the escape of ammonia. Steam was then let through for about 3 minutes [until the amount of liquid in the receiving conical flask was twice that of what it was at the beginning of the distillation]. This was then titrated with 0.01N HCl to end point and the titre value was recorded. The crude protein content was calculated from the titre value using the following relation

$$\% \text{ nitrogen} = \frac{TVXnaX \cdot 0.014 \cdot XvolX \cdot 100}{massofsampleX \cdot m'sofaliquot}$$

$$\% \text{ of nitrogen} \times CF = \% \text{ crude protein}$$

Crude lipid determination

The analysis was carried out using the method of Oyenuga (1978).

Two grams of the dried sample was placed into a thimble with the opening plugged with cotton wool as an alternative, the sample was wrapped in a filter paper. The thimble was then introduced into a barrel of the extractor and a 500L round bottom flask of known mass (156.1g) was filled with petroleum ether up to $\frac{3}{4}$ its volume. The flask was then heated at 50°C for 6 hours of the total extraction after which fat was evaporated. The residue [crude lipid] left in the flask was then weighed to know the crude lipid content.

$$\% \text{ lipid} = \frac{massof \text{ hecrudelipid}}{massofsample} \times 100$$

Ash content determination

The crucible was first washed, dried in an oven at 180°C for 30minutes cooled and then weighed (M₀ [g]). Two grams of sample was placed in the crucible and weighed, (M₁ [g]) then the crucible was transferred into the muffle furnace, whose Temperature was set at 600°C and allowed to stay for 3 hours, until the content became white after which the crucible was cool in a desiccators and weighed (M₂ [g]).

The percentage ash content was then calculated using the relation below;

$$\% \text{ ash content} = \frac{\text{mass ofash [g]}}{\text{massofsam ple [g]}} \times 100$$
$$= \frac{M_2 - M_0}{M_1 - M_0} \times 100$$

Moisture content determination

The method of Oyenuga (1978) was employed as follows:

The sample weighed (M₁) was heated in an oven at 105°C for 24hrs and cooled in a desiccator for 15 minutes then weighed, (M₂). The crucible was then returned into the oven and weighed after 3 hrs for as many times as possible until a constant value was obtained.

$$\% \text{ moisture content} = \frac{\text{lossinmassbydryingX} \cdot 100}{massofsample}$$

Crude fibre determination

100 ml of 1.25% H₂SO₄ digestion mixture was added to Two grams of sample in 250 ml conical flask, followed by occasional shaking for 30mins. The mixture was filtered through a filter paper. 100 ml of boiling water, 100mls 1.25% NaOH 20 mills of ethanol and 20ml of petroleum ether were used to wash it. The residue was dried at 100°C to constant mass in the oven, followed by ashing at 550°C to burn off the crude fibre content. The ash obtained was weighed and the crude fibre content was determined from the decrease in mass.

$$\% \text{ crude fibre} = \text{mass of crude fibre in digested cake divided by mass of sample used, multiplied by a hundred.}$$

length of giant sized castor seed species (2cm long) as reported by Weis (1971) was more than two times the length of those found in Sokoto state (0.83cm). It was however in the range of those reported from Ghana 0.9 to 1.0cm in length (Mensah and Orchan, 2005). The colours of black and light brown with white stripes of castor seeds found in Sokoto were similar to those found elsewhere (Willcox, 2000), (Weis, 1971) and (Mensah and Orchan, 2005). The number of seeds per pod in Sokoto (3) was very much within the range (3 to 4) reported by Weis (1991). The average number of pods per bunch (43) and bunch per plant (6) in Sokoto was comparable to those from other regions such as India as reported by Weis (1991). The value also suggests good yield for castor plants from Sokoto.

The results presented in table 3 illustrate the content of various mineral contents; sodium (Na), magnesium (Mg), phosphorous and calcium. Potassium shows high percentage composition of 1850 mg/ml which is greater than that of phosphorous, calcium and magnesium. The above mentioned nutrients contributed a lot in the development and growth of the plant *Ricinus communis* and could be a good source of fertilizer. The results obtained from the proximate analysis of the seed cake of castor plant indicated a high percentage of lipid content of 55.5% the high percentage lipid content showed that the plant (*Ricinus communis*) has the potential of yielding considerable amount of oil.

Carbohydrate content of the seed cake yielded (28.39%) showed that the cake may be a very good source of energy (Carbohydrate) for both man and animals if detoxified. The crude protein (11.11) also showed that the cake may be a very good source of protein. The ash content 3.5 shows highlight of the inorganic material present. It could be seen that high, percentage of ash content means high content of mineral while low content of minerals is indicated by the low percentage of ash content was indicated by the low percentage of ash content (Black, 1995).

V. Conclusion

R. communis plants in Sokoto has good yield of seeds with an average of 720 with a mass of 129.6g per plant. The study suggests that the seed has good source of carbohydrate for human and animal consumption. It is also a good source of fertilizer and a possible good oil yield that has wide variety of uses.

VI. Recommendations

In view of the high potentials of *R. Communis* as a useful crop in Sokoto, further study is recommended on the safety of the use of castor products.

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